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Development of new metal-thiosemicarbazone complexes using visual screening methods and *in silico* models

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ABSTRACT

The stability constants (log β_{11}) of forty-two new metal-thiosemicarbazone complexes were predicted based on the results of the quantitative structureproperty relationship (QSPR). The QSPR models were developed from 88 $\log \beta_{11}$ values of experimental complexes by using the multivariate linear regression (QSPR_{MLR}) and artificial neural network (QSPR_{ANN}). Four descriptors such as xch9, xv0, core-core repulsion and cosmo area were found out in the best of the linear model QSPR_{MLR} which was harshly evaluated by the statistical values: $R^{2}_{train} = 0.864$, $Q^{2}_{LOO} = 0.840$, SE = 0.711, F_{stat} = 131,355 and PRESS = 49.31. Furthermore, the artificial neural network model QSPR_{ANN} with architecture I(4)-HL(5)-O(1) was discovered with the same variables of the QSPR_{MLR} model that the statistical results were extremely impressive as $R^{2}_{train} =$ 0.970, Q^2_{CV} = 0.984 and Q^2_{test} = 0.974. Also, both of the QSPR models were externally validated on the data set of 18 log_β11 values of independently experimental complexes. As a consequence, the results from the QSPR models could be used to calculate the stability constants of other new metalthiosemicarbazones.

Introduction

The quantitative structure and property relationship (QSPR) method is known as the popular *in silico* method and it is also used widely in many fields for predicting properties of chemical compounds based on the relationships between the structural characteristics and the properties [1]. The QSPR modeling techniques are developed from a quantitative structure and activity relationship (QSAR) approach in which the properties of the model are replaced by activity and it was first introduced by Crum Brown and Fraser in 1868 [1]. In the 1940s, the appearance of chemical graph theory

and the publications of Wiener and Platt's research helped the development of QSPR modeling [1]. According to statistics of OECD, the number of published works related to QSPR models was about 11,000 projects [1]. Nowadays, the QSPR method is widely used and is seen as an effective method for finding new compounds. The QSAR/QSPR model should meet the requirements of the OECD principles [2] and the model acceptance criteria of Golbraikh and Tropsha's [1]. The basic equation of the QSPR method can be expressed mathematically as follows [1]:

Response (property) = *f*(descriptors)

(1)

Normally, there are two popular methods to establish QSPR models, they are linear regression (MLR, PLS, PCR) and machine learning methods (SVR, ANN) [1]. In this study, we use two approaches to build the QSPR models which are MLR and ANN. In addition, the diverse structure and easy complexation with many metal ions of thiosemicarbazone derivatives led to its wide applications in many fields [3]. This is the reason why thiosemicarbazone derivatives and their complexes are popularly studied in practice. In this work, we modeled the dataset of the logarithm of stability constants (log β_{11}) of the complexes (M:L) between thiosemicarbazone ligands with some metal ions (M = Mo⁶⁺, Cu²⁺, Co²⁺, Ni²⁺, Mn²⁺, Mg²⁺, Zn²⁺, Cd²⁺, Pb²⁺, Cd^{2+}) in aqueous solution. The log β_{11} values were mined from an experimental published database (table 1). The

QSPR models were formed by using multiple linear regression (QSPR_{MLR}) and the artificial neural network (QSPR_{ANN}). Besides, a new series of thiosemicarbazone ligands and complexes were designed and calculated the stability constant by the results of the developed QSPR models.

Experimental

Data selection

b)

This study selects the complex of the ML type that formed between a metal ion (M) and a thiosemicarbazone ligand (L). The structure of the selected complexes is shown in fig 1.





Figure 1: The structure of the thiosemicarbazone (a) and the metal-thiosemicarbazone complex (b)

The data mining is the first step in the QSPR modeling research. Firstly, a big data was mined from prestigious data spring, then the k-means method are approached to divide it into several data clusters [1]. In this study, a

a)

data set including the 88 $\log\beta_{11}$ values of the complexes between metal ions and the ligand thiosemicarbazone was used to build QSPR modeling in table 1.

				(logβ	11,max) value	2S			
No		Thios	semicarba	zone ligand	Metal	Number of	log	log	Def
	R ₁	R2	Rз	R4	ions	complexes, <i>n</i>	IOGB11,min	IOG <i>B</i> 11,max	Rel.
1	11	11	11		N1-6+	1	CDDCE	CDDCE	۲ <i>4</i> ٦

Table 1: The 88 stability constants of complexes (n) in experimental dataset with minimal (log $\beta_{11,min}$) and maximal

INU	R1	R_2	Rз	R4	ions	complexes, <i>n</i>	109p11,min	iogp11,max	Rel.
1	Н	Н	Н	-C ₆ H ₂ (OCH ₃) ₂ OH	M0 ⁶⁺	1	6.3365	6.3365	[4]
2	Н	Н	Н	-C ₆ H ₂ (OCH ₃) ₂ OH	Cu ²⁺	1	6.2355	6.2355	[5]
3	Н	$-C_6H_5$	-CH3	-CCH3=N-OH	Cu ²⁺	2	7.4183	7.7559	[6]
4	Н	Н	-CH₃	-CH=N-NHC ₆ H ₅	Co ²⁺	4	9.9000	10.220	[7,8]
5	Н	Н	-CH3	-CH=N-NHC ₆ H ₅	Ni ²⁺	4	10.790	11.080	[7,8]
6	Н	Н	-CH₃	-CH=N-NHC ₆ H ₅	Mn ²⁺	3	9.600	9.870	[8]
7	Н	-CH3	-CH3	-CH=N-NHC ₆ H ₅	Cu ²⁺	3	11.980	12.300	[8]
8	Н	-CH₃	-CH₃	-CH=N-NHC ₆ H ₅	Ni ²⁺	3	10.910	11.210	[8]
9	Н	Н	Н	-C10H6OH	Mg ²⁺	4	3.250	3.400	[9]
10	Н	Н	Н	-C10H6OH	Mn ²⁺	1	5.670	5.670	[9]
11	Н	Н	Н	-C10H6OH	Cd ²⁺	2	6.470	6.560	[9]
12	Н	Н	Н	-C10H6OH	Pb ²⁺	2	6.570	6.680	[9]
13	Н	Н	Н	-C10H6OH	Zn ²⁺	1	7.170	7.170	[9]
14	Н	Н	Н	-C10H6OH	Co ²⁺	4	7.890	9.000	[9]
15	Н	Н	Н	-C10H6OH	Ni ²⁺	4	8.370	9.600	[9]
16	Н	Н	Н	-C10H6OH	Cu ²⁺	4	8.680	9.780	[9]

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17	Н	Н	-	-C9H8NO	Ni ²⁺	8	7.709	8.500	[10]
18	Н	Н	-	-C9H8NO	Pb ²⁺	7	7.307	8.109	[10]
19	Н	Н	-	-C9H8NO	Co ²⁺	8	7.251	8.340	[10]
20	Н	Н	-	-C9H8NO	Zn ²⁺	8	7.039	8.160	[10]
21	Н	Н	-	-C9H8NO	Cd ²⁺	6	6.611	7.599	[10]
22	Н	Н	-	-C9H8NO	Mn ²⁺	8	5.439	6.230	[10]

Descriptors

Molecular descriptors are known as the variables in the equations of the QSPR models and they consist of physicochemical, topological indicator and quantum parameters related chemical structures. The descriptors of the metal-thiosemicarbazone complexes were generated using QSARIS package [11] and calculated by using the semi-empirical quantum method with new version PM7 and PM7/sparkle on the MoPac2016 system [12].

QSPR modeling methods

MLR method

In the multivariate linear regression (MLR) modeling methods, the equation of the $QSPR_{MLR}$ model is described according to the following equation [1,13]:

$$Y = b_0 + \sum_{j=1}^{k} b_j X_j$$
 (2)

where b_0 , is the intercept of the model, b_j is the regression coefficients and k is number of explanatory variables in the equation.

The MLR analysis is performed by stepwise regression technique on the Regress system [13] and MS-EXCEL [14] and the cross-validation (CV) method for the QSPR models is carried out by the leave-one-out process (LOO) [1,13].

ANN method

In general, an artificial neural network (ANN) model includes an input layer, one or more hidden layer, and an output layer. Neurons in each of the layers are called nodes that interconnect with each other and receive linked weights. The typical ANN architecture used in many studies is multi-layer perceptron (MLP) for the formation of the models [15].

In this study, the MLP-ANN type is used with an error back-propagation algorithm [15]. The architecture consists of three layers I(k)-HL(m)-O(n). The input layer (k) contains the variables of the MLR model, an output layer (n) is the stability constant log β_{11} and the number of hidden neurons (m) is specified by neurons on the

input and output layer. To find out the best ANN architecture for QSPR_{ANN} model, the process is operate within two steps. Firstly, the *m* values of hidden neurons are examined by using Neural Designer tools [16], then a data set is used to externally validate the QSPR_{ANN} model from the results of surveyed models. The evaluation of the second step are run on the Matlab 2016a with Neural Network tool (nntool) toolbox [17].

In addition, two basic transfer functions in the neural network such as the hyperbolic sigmoid tangent and log-sigmoid transfer function are used in the training process of ANN models. The transfer functions are represented mathematically as follows [15-17]:

$$a = \tan sig(n) = \frac{2}{\left(1 + e^{-2n}\right)^{-1}}$$
(3)

$$a = \log sig(n) = \frac{1}{1 + e^{-n}}$$
 (4)

Model Validation

The validation of the QSPR_{MLR} model was appreciated by the statistical parameters like explained variance (R²_{adj}), coefficient of determination (R²_{train}), standard error (SE), Fischer's value (F_{stat}), and predicted residual sum of squares (PRESS) [1]. The MLR models were trained with internal validation by LOO statistics (Q²_{LOO}) and validated externally on another data set using the statistic Q²_{ex}.

The QSPR_{ANN} models were controlled by three coefficients of determination such as R^2_{train} for training set, Q^2_{test} for test set and Q^2_{CV} for cross-validation set. The ANN model is trained till the mean square error (*MSE*) value is minimized followed by a difference of the output (*o*) and real (*r*) values [17]. It is described as followed [15-17]:

$$MSE = \frac{1}{n} \sum_{1}^{n} (r_i - o_i)^2$$
(5)

This work uses the average absolute values of the relative errors MARE (%) where ARE (%) is the absolute value of the relative errors to compare the quality of the models [18]. Furthermore, to find out the variable contributions in the models, The average contribution https://doi.org/10.51316/jca.2021.096

percentage $(MPx_{k,i})$ is used and it is determined according to formula (6) [18]:

$$MPx_{k,i}, \% = \frac{1}{N} \sum_{m=1}^{N} \frac{100. \left| b_{k,i} \cdot x_{m,i} \right|}{\sum_{j}^{k} \left| b_{k,j} \cdot x_{m,j} \right|}$$
(6)

where *N* is the number of observations; *m* is the number of substances used to calculate $Px_{k,i}$ value; $b_{k,i}$ are the parameters of the model.

Results and discussion

QSPR_{MLR} models

As above-mentioned, the data set for the construction of QSPR_{MLR} including the 88 stability constants values of complexes are divided into a training set of 71 observations (80 %) and a test set of 17 observations (20 %) [1]. The training process of QSPR_{MLR} models is based on the criteria of statistical values to evaluate the quality of models [1]. The results of QSPR_{MLR} models and the statistical values are shown in table 2.

k	Variables	SE	$R^2_{\rm train}$	R^2_{adj}	Q^2 LOO	F _{stat}	PRESS
1	<i>X</i> 1	1.094	0.666	0.662	0.649	171.2253	108.02
2	<i>x</i> ₁ / <i>x</i> ₂	0.885	0.783	0.778	0.770	153.8012	70.74
3	<i>X</i> 1/ <i>X</i> 2/ <i>X</i> 3	0.814	0.819	0.813	0.796	126.7552	62.67
4	X1/X2/X3/X4	0.711	0.864	0.857	0.840	131.3551	49.31
5	X1/X2/X3/X4/X5	0.660	0.884	0.877	0.861	924.8164	42.88
Notation	of molecular descripte	ors					
xch9		<i>X</i> 1	(Cosmo area		<i>X</i> 4	
xv0		<i>X</i> ₂		xvch5		X5	
core-core	e repulsion	Хз					

Table 2: Selected models QSPR_{MLR} (k of 1 to 5) and statistical values

The number of the variables (*k*) for the best QSPR model is selected on basis of the changing the R^2_{train} , Q^2_{LOO} , F_{stat} , and *PRESS* values for meeting statistical requirements.^[1,13] Moreover, the variables from x_1 to x_5 were closely monitored on the basis of the p-value (< 0.05) and t-student characterized by the variables [1,13]. The results of fig 2a show that when the *k* values

r

increase to 5, the QSPR model gets the best statistical values; however, in this case, we choose the *k* value of 4 because when the k value increase from 4 to 5, the statistical values increase insignificantly and this is unnecessary for the building of models. So the best QSPR_{MLR} model was selected with the statistical values as follows (7):

$$\log \beta_{11} = 10.29 - 261.86 \cdot x_1 + 0.763 \cdot x_2 + 0.00038 \cdot x_3 - 0.053 \cdot x_4$$

$$p = 88; R^2_{\text{train}} = 0.864; Q^2_{\text{LOO}} = 0.840; SE = 0.711; Fstat = 131.3551; PRESS = 49.31$$

(7)



Figure 2: (a) Change tendency line of values SE, R^2_{train} and Q^2_{LOO} according to k values; (b) Correlation of experimental vs. predicted values $\log \beta_{11}$ of the training compounds using the QSPR_{MLR} model (with k = 4)

https://doi.org/10.51316/jca.2021.096 305 As a consequence, this data set used to build the MLR is completely agreeable and the $QSPR_{MLR}$ model has good predictability for complex groups. Therefore, the model can be used to predict new complexes of the same type group by using the Applicability Domain (AD) and Outliers estimates [1,2].

The investigation uses three models in the vicinity of selected *k* values to validate the role of descriptors in the model according to the *GMPx_i* values (*GMPx_i* is the average value of *MPx_{k,i}*). The results in Table 3 show that the contribution of the variables in the order of xv0 (x_2) > Cosmo area (x_4) > core-core repulsion (x_3) > xch9 (x_1) corresponds to the values of

37.2860, 31.4096, 18.9904, and 11.7660. The xv0 parameter (xv0 = ${}^{0}\chi^{v}$ in Kier & Hall notation), namely Chi valence zero, which is a zero-order valence connectivity index computed over all vertices (atoms) in the entire molecules [11].

Cosmo area and core-core repulsion are the quantum parameters, in which Cosmo area is the total area of molecules [12]. The xch9 parameter, namely Chi chain 9, is the simple 9th-order chain chi index and the Chi index is the sum of the 9th-order chain of subgraph [11]. The important parameters are selected to design new complexes and predict the stability constant of these complexes.

Table 3: Statistical values and variables,	MPxki and GMPxi contribution ir	n models $QSPR_{MLR}$ with k of 4 to 6

Statistical values		QSPR _{MLR}			$MPx_{k,i}$, %		
and variables	<i>k</i> = 3	<i>k</i> = 4	<i>k</i> = 5	<i>k</i> = 3	<i>k</i> = 4	<i>k</i> = 5	GMPXi, %
R^2 train	0.819	0.864	0.884	-	_	-	_
R^2 adj	0.813	0.857	0.877	_	-	-	_
Q^{2}_{LOO}	0.796	0.840	0.861	_	_	-	-
RMSE	0.814	0.711	0.660	_	-	-	_
constant	-2.603	10.29	19.13	_	_	-	-
<i>X</i> 1	-197.09	-261.86	-315.07	16.1301	10.3839	8.7840	11.7660
<i>X</i> 2	-0.695	0.763	0.931	57.0639	29.5652	25.2288	37.2860
<i>X</i> 3	0.000284	0.000383	0.000423	26.8059	17.0347	13.1308	18.9904
<i>X</i> 4	-	-0.05344	-0.09108	_	43.0162	51.2125	31.4096
X 5	_	_	-2.100	_	_	1.6439	0.54798

QSPRANN models

The ANN model is built on the same variables of the MLR model, so the model is developed upon 4 variables of the QSPR_{MLR} model. The data set is split into a training set of 60 observations (70 %), a test set of 14 observations (15 %), and a cross-validation set of 14 observations (15 %) [17]. The architecture of the

neural network is I(4)-HL(m)-O(1), in which four neurons of the input layer I(4) are xv0, Cosmo area, core-core repulsion and xch9; one neuron of the output layer O(1) is the log β_{11} values and the number of hidden layer (m) is scanned and the results of the m neurons are given in table 4.

Table 4: The developed	QSPRANN model	I(4)-HL(m)-O(1)	with statistical	parameters
------------------------	---------------	-----------------	------------------	------------

No	OSPRANN model	R^{2}_{train}	O^{2}_{tort}	O^2 CV	Training	Test	Validation	Transfer Function
110	Q3I HANN HIOGEI	in train		Q CV	error	Error	Error	indifisient directori
1	I(4)-HL(6)-O(1)	0.9840	0.9765	0.9853	0.0503	0.1216	0.0664	hyperbolic tangent
2	I(4)-HL(8)-O(1)	0.9753	0.9615	0.9802	0.0775	0.1707	0.0829	hyperbolic tangent
3	I(4)-HL(6)-O(1)	0.9640	0.9800	0.9830	0.1129	0.0981	0.0672	log-sigmoid
4	I(4)-HL(5)-O(1)	0.9700	0.9739	0.9836	0.0951	0.1129	0.0616	hyperbolic tangent
5	I(4)-HL(8)-O(1)	0.9668	0.9697	0.9848	0.1040	0.1245	0.0655	log-sigmoid



Figure 3: (a) The architecture of neural network I(4)-HL(5)-O(1); (b) The correlation between experimental vs. predicted values of external data set of QSPR models

Using a data set of 18 samples to find out the best QSPR_{ANN} model by external validation technique, the results show that the QSPR_{ANN} model I(4)-HL(5)-O(1) (bolded in table 5) is the best predictability with the Q^2_{ex} value of 0.8936 as in fig 3b. As a consequence, the hyperbolic tangent transfer function is used for network training and the optimum parameters of ANN model such as the momentum constant of 0.05, the learning rate of 0.01, and the convergent goal of 10⁻¹⁰.

External Validation

To build a good model, it is essential to perform the external evaluation on a data set that is independent of the training data set.^[1] This work used the external data of 18 observations from the experimental studies. The full information of 18 samples and the validated results are presented in table 5.

As observed in table 5 and fig 3b, the data analysis from the calculated results indicate the neural network and linear regression models express the correlation between the predicted values and the experimental values with the $Q^2_{\rm ex}$ values of 0.8936 and 0.8193, respectively. Hence, the predictability of the two models is extremely positive [1].

As a further matter, the *MARE* values of QSPR_{MLR} and QSPR_{ANN} I(4)-HL(5)-O(1) models are 9.578 % and 5.318 %, respectively. The results point out that the ANN model has better predictability than the MLR model. The predicted $\log \beta_{11,cal}$ values of ANN model approximate to the experimental $\log \beta_{11,exp}$ values. Furthermore, using the one-way ANOVA method to evaluate the difference between the experimental and predictive values of both models; accordingly, the differences between the QSPR models are negligible (F = 0.1402 < F_{0.05} = 3.1788).

	Thios	emicarba	zone ligand	Metal	loge	logβ	$logeta_{11,cal}$		
R ₁	R ₂	Rз	R4	ions	109P11,exp	QSPR _{MLR}	QSPRann	rei.	
Н	Н	Н	-C ₁₃ H ₁₆ NO ₃	Zn ²⁺	12.400	10.025	11.309	[19]	
Н	Н	Н	-C ₆ H ₃ (OH)OCH ₃	Cd ²⁺	7.340	7.693	7.502	[20]	
Н	Н	Н	-C ₆ H ₃ (OH)OCH ₃	Co ²⁺	7.940	7.354	8.245	[20]	
Н	Н	Н	-C ₆ H ₃ (OH)OCH ₃	Ni ²⁺	8.370	7.827	8.886	[20]	
Н	Н	Н	-C ₆ H ₃ (OH)OCH ₃	Cu ²⁺	9.030	8.276	9.464	[20]	
Н	Н	Н	-C ₁₀ H ₆ OH	Mg ²⁺	3.400	3.967	3.374	[9]	
Н	Н	Н	-C10H6OH	Mn ²⁺	5.670	6.510	5.600	[9]	
Н	Н	Н	-C ₁₀ H ₆ OH	Cd ²⁺	6.560	7.933	7.664	[9]	
Н	Н	Н	-C10H6OH	Zn ²⁺	7.170	7.454	7.136	[9]	
Н	Н	Н	-C ₁₀ H ₆ OH	Co ²⁺	8.130	7.792	8.376	[9]	
Н	Н	Н	-C10H6OH	Ni ²⁺	9.070	8.293	8.959	[9]	
Н	Н	Н	-C ₁₀ H ₆ OH	Cu ²⁺	9.290	8.591	9.244	[9]	
Н	Н	-	-C9H8NO	Cu ²⁺	9.06	8.371	9.104	[10]	
Н	Н	Н	$-C_6H_4NH_2$	Cu ²⁺	10.570	10.838	9.017	[21]	
Н	Н	Н	$-C_6H_4NH_2$	Zn ²⁺	8.740	9.941	9.211	[21]	

Table 5: The experimental log $\beta_{11,exp}$ and external predicted log $\beta_{11,cal}$ values from the QSPR models

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Н	Н	Н	-C ₆ H ₄ NO ₂	La ³⁺	10.840	9.823	12.226	[21]
Н	Н	Н	$-C_6H_4NO_2$	Pr ³⁺	11.040	9.788	12.222	[21]
Н	Н	Н	-C6H4NO2	Cd ²⁺	10.630	11.123	10.439	[22]
					MARE, %	9.578	5.318	

Development of new complexes

Using four descriptors such as xv0, Cosmo area, corecore repulsion, and xch9 of the resulted models and the AD and Outliers rules to develop the new complexes, the phenothiazine and carbazole derivatives are selected to design new thiosemicarbazone and the complexes among the new ligands with several popular metal ions as Ag⁺, Cd²⁺, Cu²⁺, Ni²⁺, and Zn²⁺. These derivatives have been synthesized in experimental researches [23-26]. The new thiosemicarbazones are formed by attaching phenothiazine and carbazole groups at the R_4 site while the remaining positions as R_1 , R_2 , and R_3 of the thiosemicarbazones are hydrogen atoms.

A series of new complexes are designed and screened carefully by embedding their descriptors into the spatial data of the training data set [1,2]. The results, including 42 new complexes, meet the standard of AD and Outliers rules via D-Cook values [1,2], and the stability constant ($\log \beta_{11,new}$) are predicted by the two built QSPR_{MLR} and QSPR_{ANN} models (table 7).

Table 6: Forty-two new metal-thiosemicarbazone co	omplexes with the	e calculated log $eta_{ ext{l1,ne}}$	w values from th	he built
QSPR	R models			

	Metal	logβ	, 11,new		Metal	logß	, 11,new
R4 SITE	ions	MLR	ANN	R4 site	ions	MLR	ANN
	Ag+	14.0992	11.8642	ONa	Ag+	11.8770	11.8863
	Cd ²⁺	13.1645	11.8726	o s N	Cd ²⁺	10.8608	11.8896
	Cu ²⁺	13.7560	11.3848	\sim	Cu ²⁺	11.5941	11.8514
0 ₂ N	Ni ²⁺	13.3203	11.1200		Ni ²⁺	10.9582	11.8453
	Zn ²⁺	12.8330	11.7337	Br	Zn ²⁺	10.2510	11.8810
	Ag+	6.1956	5.0159		Cd ²⁺	14.2768	11.8900
	Cd ²⁺	5.1529	4.9896	Ń	Ni ²⁺	14.3241	11.8790
	Cu ²⁺	6.1128	5.1800		Zn ²⁺	13.8725	11.8877
ССССССССССССССССССССССССССССССССССССС	Ni ²⁺	5.0871	5.4205	Br	Ni ²⁺	6.8250	8.1519
	Zn ²⁺	4.9812	5.1407	H H H H H H H H H H H H H H H H H H H	Zn ²⁺	9.0312	10.4932
HO	Ag ⁺	5.7760	6.3457		Ag ⁺	6.3136	6.3913
	Zn ²⁺	4.7133	5.6821		Zn ²⁺	5.2077	5.4452
HO	Cd ²⁺	6.3349	5.9627		Cd ²⁺	6.7341	4.9664
HO	Cd ²⁺	5.7434	6.4616	HO	Cd ² +	5.8818	6.7292
HO	Cd ²⁺	6.2653	5.7119	H H H	Zn ²⁺	8.8060	7.1983

HO	Cd ²⁺	10.8266	11.7292		Ag+	9.0288	7.5373
° ↓ ↓	Zn ²⁺	10.5792	10.4042		Zn ²⁺	7.9959	7.8082
6 Th o	Ag+	9.2069	8.8589		Ag ⁺	10.5251	10.2175
O H	Zn ²⁺	8.0592	8.9143		Zn ²⁺	9.2111	9.0465
	Ag+	9.3004	9.3045		Ag ⁺	6.9281	7.9893
OH OH	Zn ²⁺	7.8424	7.7793	н	Zn ²⁺	5.8215	7.0230

Furthermore, the single-factor ANOVA method is used to compare the predicted $\log\beta_{11,\text{new}}$ values from the resulted QSPR_{MLR} and QSPR_{ANN} models. It indicates that there is no difference between the two models ($F = 0.1930 < F_{0.05} = 4.0195$).

Conclusion

In this investigation, the QSPR models based on the multivariate linear regression and artificial neural network methods were successfully formed by using the dataset of structural descriptors and the stability constant values of metal-thiosemicarbazone complexes. The structures of the complexes were optimized by semi-empirical guantum mechanics with the new version PM7. Also, the QSPR models were completely validated upon the statistical values as R^{2}_{train} , Q^{2}_{LOO} , Q^{2}_{CV} , MARE (%), and ANOVA methods. The resulted models are acknowledged as being novel models for the prediction of forty-two new designing thiosemicarbazone derivatives. Furthermore, the results of QSPR models can be effective for discovering new other complexes that can be further applied in essential fields such as analytical chemistry, environment monitoring, and drug design in pharmacology.

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